Code No. 3005/BL

FACULTY OF ENGINEERING & INFORMATICS

B.E. I – Year (Backlog) Examination, May / June 2017

Subject: Engineering Chemistry

Max.Marks: 75

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Note: Answer all questions from Part A and any five questions from Part B.

Time: 3 Hours

PART – A (25 Marks)

1	Differentiate between electrolytic and electrochemical cell.	3
2	Write the chemical reaction involved in H ₂ -O ₂ fuel cell	2
3	Explain Waterline corrosion.	2
4	What is Break-point chlorination? Give its significance	3
5	What is meant by Degree of Polymerization?	2
6	Mention a few applications of Composites.	2
7	What are the characteristics of a good propellant?	2
8	What is octane number? What is its significance	3
9	What is saponification number? What is its significance?	3
10	What are the principles of Green Chemistry?	3
	PART – B (5x10 = 50 Marks)	
11	a) Derive Nernst equation for the calculation of cell EMF and give its applications.	6
	b) What is the EMF of the following cell at 25° C.	
	Zn _(s) /Zn ²⁺ (0.2M)//Ag ⁺ (0.002M)/Ag _(s) .	
	The standard EMF of the cell is 1.54v	4
12	a) What is Electrochemical Corrosion? Describe the mechanism of Electrochemical	
	corrosion.	5
	b) Define Alkalinity of water. How is it determined?	5
13	a) Describe the method of preparation, properties and application of the following:i) Teflon	
	ii) Perlon-U	6
	 b) What is Conducting Polymer? Explain the mechanism of conduction in Polyacetylene. 	4

14	a)	What is meant by Cracking of Petroleum? Explain moving bed catalytic cracking	ıg
		method.	6
	b)	A sample of coal contains 60% C, 33%O, 6%H, 0.5%S, 0.2% N & remaining as	h.
		Calculate the Gross and Net calorific value of coal.	4
15	a)	Draw a neat labeled Phase Diagram of Water system and explain Areas, Curves &	
		Triple point in it.	6
	b)	Write a note on Extreme Pressure Lubrication.	4
16	a)	Differentiate between Anodic & Cathodic coatings.	4
	b)	Write a note on potentiometric titrations.	6
17	a)	Explain Fractionation of Petroleum Crude with the help of a neat diagram	5
	b)	Define Liquid Crystals. Explain about the Thermotropic Liquid crystals.	5