

FACULTY OF ENGINEERING & INFORMATICS**B.E. I-Semester (Suppl.) Examination, June / July 2017****Subject : Engineering Chemistry-I****Time : 3 hours****Max. Marks : 70****Note: Answer all questions from Part-A. Answer any FIVE questions from Part-B.****PART – A (20 Marks)**

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| 1 | State and explain First law of thermodynamics. | 2 |
| 2 | State Carnot theorem. | 2 |
| 3 | What is meant by the term eutectic? | 2 |
| 4 | State phase rule. | 2 |
| 5 | Calculate the carbonate and non-carbonate hardness of a sample of water in ppm containing : $\text{Ca}(\text{HCO}_3)_2 = 8.1 \text{ mg.L}^{-1}$; $\text{Mg}(\text{HCO}_3)_2 = 7.3 \text{ mg.L}^{-1}$; $\text{MgCl}_2 = 9.5 \text{ mg.L}^{-1}$; $\text{CaSO}_4 = 13.6 \text{ mg.L}^{-1}$. | 2 |
| 6 | Define the terms i) Scale and ii) sludge | 2 |
| 7 | Give one example each for Addition and Condensation polymers. | 2 |
| 8 | Write the structures of poly-acetylene and poly-aniline. | 2 |
| 9 | Define the terms i) Saponification number and ii) acid value | 2 |
| 10 | Explain the property of RUL in refractories. | 2 |

PART – B (50 Marks)

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| 11 a) | Calculate the maximum work done when 2 moles of an ideal gas expand isothermally and reversibly from a volume of 10 litres to a volume of 20 litres at 298k. | 5 |
| b) | Explain the criteria for spontaneity of a process in terms of entropy and free energy. | 5 |
| 12 a) | What do you understand by the reduced phase rule equation? Discuss the use of the phase rule in Pattinson's process of desilverization of lead. | 5 |
| b) | Define the terms i) Phase ii) component iii) degrees of freedom. | 5 |
| 13 a) | Explain the procedure for the determination of Alkalinity of water. | 5 |
| b) | Discuss the concept of break point chlorination. | 5 |
| 14 a) | Explain the preparation, properties and applications of Nylon-6,6. | 5 |
| b) | Differentiate between thermoplastic and thermosetting polymers. | 5 |
| 15 a) | Classify lubricants and give one example each for various type of lubricants. | 5 |
| b) | Write a note on the following properties of Refractories.
i) Refractoriness ii) Thermal spalling | 5 |
| 16 a) | Derive an expression for the efficiency of heat engine by using carnot cycle. | 6 |
| b) | Discuss the ion-exchange method of softening hard water. | 4 |
| 17 a) | Write a note on intrinsic conducting polymers | 5 |
| b) | Explain the terms i) viscosity index ii) glazing | 5 |