

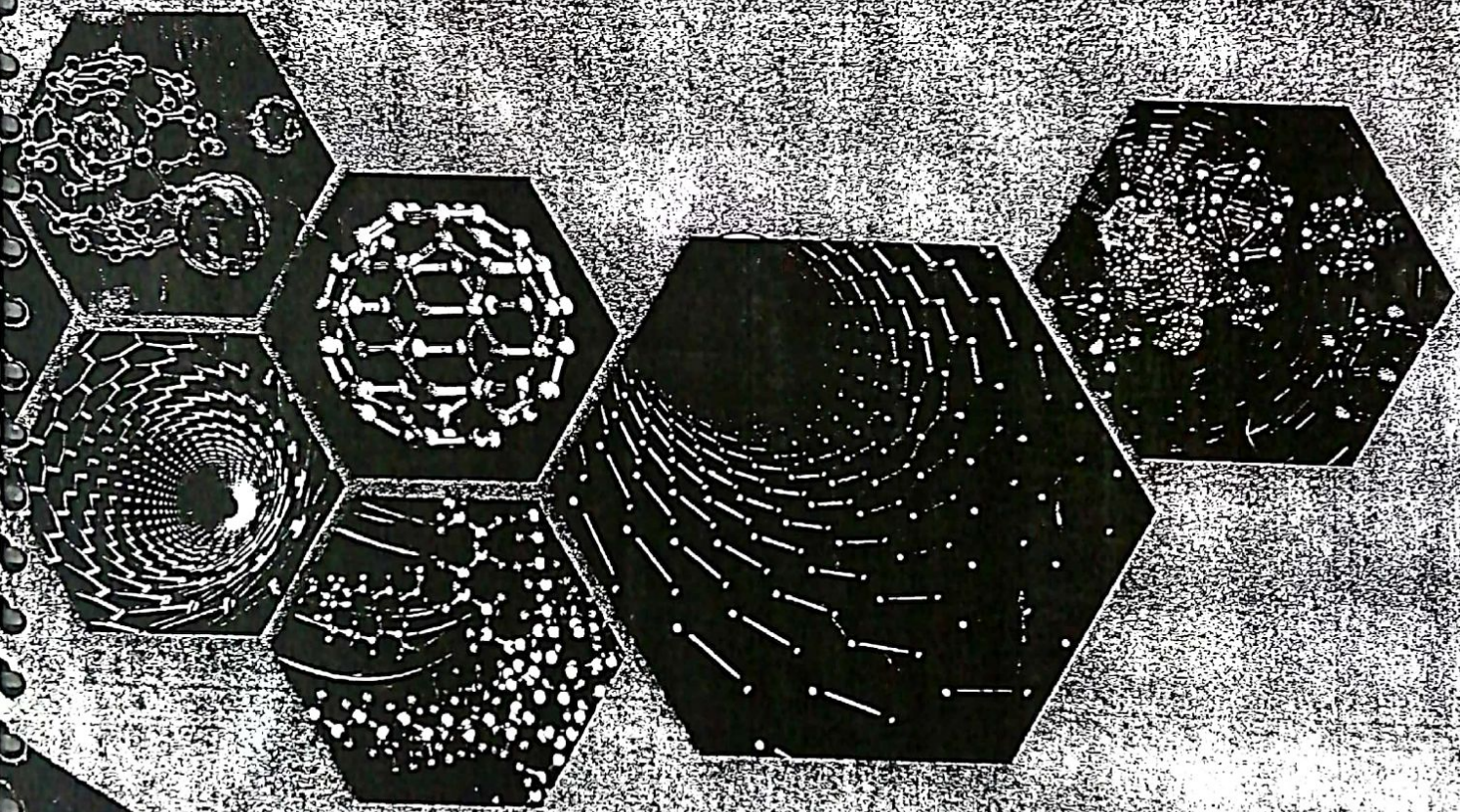


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Electrical Studies In Silver Doped Gallium Oxide Glasses

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ABSTRACT: The glasses of composition $x\text{AgI}-(100-x)[0.02\text{Ga}_2\text{O}_3-0.49\text{Ag}_2\text{O}-0.49\text{B}_2\text{O}_3]$ where $x=5$ to 30 in steps of 5 weight% are prepared by melt quenching technique. XRD, FTIR and DSC investigations are carried out on all glasses to understand physical characteristics of the prepared glasses. Electrical characterization is done in terms of DC and AC conductivities. DC conductivity at room temperature increased from 10^{-4} to 10^{-2} Scm^{-1} with increasing concentration of AgI. DC activation energy (E_{dc}) is found to decrease from 0.36 to 0.19 eV with increasing concentration of AgI. From the Impedance spectroscopy real and imaginary parts of impedances (Z' , Z'') plots plotted, and by using Z-View equivalent software R-C circuit parameters are obtained from Cole-Cole plots and relaxation times also calculated. The quantitative analysis of these results indicates that the electrical conductivity of these glasses is enhanced by addition of AgI.

Keywords: Electrical properties, oxide; Ga_2O_3 , Ag_2O , XRD

1. Introduction

Portable electronic devices, omnipresent nowadays, requires small-size power sources which should be more efficient, durable, reliable and safer for the environment than those available now. To meet growing and urgent needs for such power sources it is vital to carry out the research on many prospective ideas: miniaturized fuel cells, rechargeable microbatteries, etc. One of specific tasks in this research is the development of solid electrolytes (polycrystalline, amorphous or composite) for rechargeable silver microbatteries [1,2]. Though the ionic conductivity of many silver glasses, especially those with high contents of AgI, can be suitable for battery applications at moderate temperatures, some disadvantages of those glasses such as, e.g. their brittleness or low thermal stability threshold, limit the prospects of their practical use. One of possible solutions to circumvent the drawbacks of glassy electrolytes is to produce composites based on these glasses. There have been already explored several interesting approaches to prepare composites based on AgI and